

**SOUTH AUSTRALIA**  
**DEPARTMENT OF MINES AND ENERGY**



**OPEN FILE ENVELOPE NO. 7283/2**

**PEL 5 AND PEL 6, MURTA BLOCK**  
**EROMANGA AND COOPER BASINS**

**KOREE 1**  
**TEST REPORTS**

Submitted by

Santos Ltd

1990

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SCANNED

**ENVELOPE 7283/2**

**TENEMENT:** PEL 5 and PEL 6, Murta Block; Eromanga and Cooper Basins

**TENEMENT HOLDER:** Santos Ltd (operator), Vamgas Ltd, Delhi Petroleum Pty Ltd and SAGASCO Resources Ltd

**CONTENTS****SADME NO.**

**REPORT:** McDonough, R., 1991. Patchawarra (gas reservoir DST 1 flow) FWS composition  
(18 April, 1991).

**7283/2 R1**  
**Pgs 3-15**

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DEPARTMENT OF MINES AND ENERGY - SOUTH AUSTRALIA  
DEPARTMENTAL MEMORANDUM

18 April, 1991

TO : Director, Oil, Gas &amp; Coal

FROM : Richard McDonough

SUBJECT : KOREE PATCHAWARRA FWS COMPOSITION

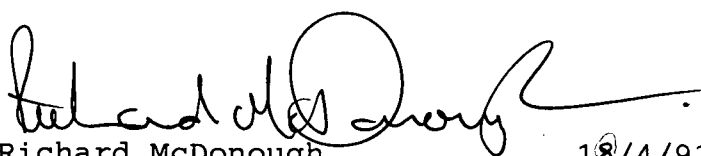
SR 27/4/7283/0

The full well stream composition of the Patchawarra formation reservoir fluid for Koree has been calculated by mathematical recombination of gas and liquid samples taken from Koree #1, DST #1, and is listed below :

Component	recombined stream (mol%)
N2	1.21
CO2	7.82
C1	64.82
C2	10.46
C3	6.76
i-C4	1.11
n-C4	2.40
i-C5	0.80
n-C5	1.04
C6	0.89
C7	1.24
C8+	1.45
total	100.00

It should be noted that, due to insufficient reserves, the well was plugged and abandoned.

This work has been done as part of the update for the 1991 Reserves Atlas.

  
Richard McDonough 18/4/91

udd\oarmc\recomb\koree.spr

## KOREE FWS COMPOSITION

FORMATION : Patchawarra

SOURCE : KOREE #1, DST #1 RECOMBINATION

CALCULATION DATE : 18 April, 1991

The full well stream composition of the Patchawarra formation reservoir fluid for Koree has been calculated by mathematical recombination of gas and liquid samples taken from Koree #1, DST #1, and is listed below :

Component	recombined stream (mol%)
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C7	1.24
C8+	1.45
total	100.00

It should be noted that, due to insufficient reserves, the well was plugged and abandoned.

### METHOD

The raw data is presented in the Appendix. This presents AMDEL analyses of HP separator gas, stock tank liquid, flash gas from stock tank liquid and the mathematically recombined HP separator liquid.

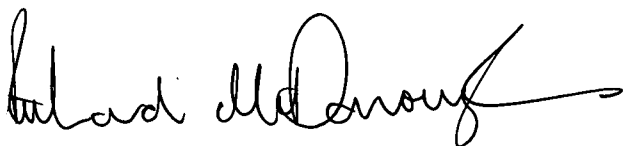
A plot of  $\log K$  vs  $(T_c)^2$  for both the flash gas/flash liquid and HP gas/recombined liquid is also shown in the Appendix. Analysis of the HP gas/recombined liquid plot shows that there is a reasonable straight line fit to the data. It was decided that no adjustment to the data was required and the recombination should proceed normally.

The separator gas rate was reported to be 3.4 mmcfd. This was converted to a molar flow rate by multiplying by the ideal gas molar volume. The stock tank liquid rate was reported to be 160 bcpd. The molar flow rate was calculated by multiplying the volume rate by the density and dividing by the calculated stock tank liquid molecular weight.

In order to perform the recombination, the HP liquid rate must be calculated. This is done by calculating the relative proportions of stock tank liquid and gas produced from the HP liquid. The stock tank rate is then divided by the ratio of liquids produced to feed volume.

The HP liquid and gas samples are then recombined according to the ratio of the molar flow rates.

The workings for this calculation are attached.



Richard McDonough

18/4/91

KOREE #1 : DST #1

=====

RAW DATA :

=====

ideal gas molar volume            379.49 ft3/lb-mole  
 g/cc ---> lb/ft3                62.43

separator gas rate        =            3.4 mmcfd  
                              =        8959.393 lb-mole/day

stock tank liquid rate =            160 bcpd  
                              =        393.67 lb-moles/day

STOCK TANK FLUID ANALYSIS

=====

From the AMDEL report, the high pressure liquid yielded 381 cc of liquid and 20.0 litres of gas at 25.1 deg C.

moles of gas produced

-----

P =        101.325 kPa  
 R =        8.314  
 V =        20 litres  
 T =        25.1 deg C            298.1 Kelvin

n =         $P*V/(R*T)$   
           =        0.817664 moles

moles of liquid produced

-----

liquid volume                                =            381 cc  
 measured liquid density @ 25.1 deg C       =            0.7135 g/cc  
    =            44.54  
 liquid mass                                   =        271.8435 grams  
 mole weight of liquid                       =            101.67  
 moles of liquid                               =        2.673707 moles

molar ratio of gas and liquid to feed

-----  
for 100 moles of HP separator liquid

gas produced = 23.42

liquid produced = 76.58

calculate HP molar flow rates

-----  
stock tank liquid rate = 393.67 lb-moles/day

HP liquid rate = 514.06 lb-moles/day

HP gas rate = 8959.39 lb-moles/day

recombination ratio

-----  
liquids ratio = 0.054263

gas ratio = 0.945737

# MATHEMATICAL RECOMBINATION

=====

	HP GAS	HP LIQ	K=Yi/Xi	recombined stream
-----				
N2	1.27	0.10	12.70	1.21
CO2	8.18	1.63	5.02	7.82
C1	68.21	5.74	11.88	64.82
C2	10.79	4.62	2.34	10.46
C3	6.67	8.28	0.81	6.76
i-C4	1.03	2.52	0.41	1.11
n-C4	2.11	7.50	0.28	2.40
i-C5	0.51	5.83	0.087	0.80
n-C5	0.56	9.45	0.059	1.04
C6	0.42	9.09	0.046	0.89
C7	0.19	19.61	0.010	1.24
C8+	0.06	25.63	0.002	1.45
-----				
total	100.00	100.00		100.00
=====				

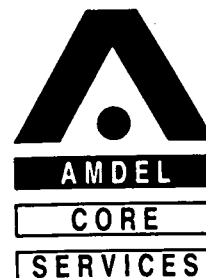
## MOLECULAR WEIGHT OF STOCK TANK LIQUID

	Xi	mol wt	Xi*MW
N2	0.00		0.00
CO2	0.00		0.00
C1	0.00		0.00
C2	0.20	30.07	0.06
C3	2.31	44.10	1.02
i-C4	2.36	58.12	1.37
n-C4	6.12	58.12	3.56
i-C5	6.69	72.15	4.83
n-C5	11.41	72.15	8.23
C6	11.32	86.18	9.76
C7	25.72	100.20	25.77
C8+	33.87	139.00	47.08
TOTAL	100.00		101.67



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# AMDEL CORE SERVICES GAS ANALYSIS

Method GL-01-01

ASTM D 1945-81 (modified)

Report # 009/354

Client: SANTOS Ltd.

Sample: KOREE 1, DST-1  
Separator: 340 psi  
Date: 10/5/90  
Cylinder #EX242

GAS	MOL %
Nitrogen	1.27
Carbon Dioxide	8.18
Methane	68.21
Ethane	10.79
Propane	6.67
I-Butane	1.03
N-Butane	2.11
I-Pentane	0.51
N-Pentane	0.56
Hexanes	0.42
Heptanes	0.19
Octanes and higher h.c	0.06
Total	100.00

( 0.00 = less than 0.01% )

Calculated Gas Density

( Air = 1 ) : 0.841

Calorific Value (15.0 deg C, 101.325 kPa)

Gross:	1230 BTU/CU Ft	45.81 MJ/CU.M
Nett:	1117 BTU/CU Ft	41.62 MJ/CU.M
Gross calorific value of water-saturated gas		45.01 MJ/CU.M
Average Molecular Weight =	24.304	

All results are calculated on the basis that only the  
measured constituents are present. This report  
relates specifically to the sample tested; it also  
relates to the entire batch insofar as the sample is truly  
representative of the batch.

Approved Signatory

*R. Tambe*

Date

16-Jun-90

AMDEL CORE SERVICES PTY LIMITED - Method GL-01-01(ASTM D 1945-81, modified)

This method conforms to ASTM D 1945-81, "Analysis of Natural Gas by Gas Chromatography"; however, this standard is quite general and permits considerable scope in the configuration of equipment and processing of results.

Tests carried out by Amdel Limited in May and June 1987 indicate that the repeatability of our analyses conforms to that specified in the standard. This being the case, we maintain that our analyses will reach the reproducibility requirement also. These precision estimates are:

Component Level, (mol. %)	Repeatability	Reproducibility
0.01 to 1	0.03	0.06
1 to 5	0.05	0.10
5 to 25	0.15	0.20
Over 25	0.30	0.60
C <sub>6</sub> and heavier fractions*	5% of amount	10% of amount

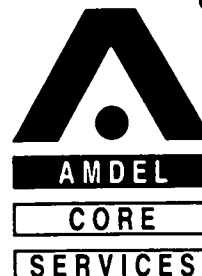
\* The standard assumes calculation of results into C<sub>6</sub> and C<sub>7(+)</sub> fractions.

The repeatability is the value below which, in 95% of cases, the difference between two single values obtained under the same conditions may be expected to lie. Reproducibility is defined similarly but refers to analyses carried out by, for example, different operators, different days or different laboratories.

For the sake of uniformity and client's convenience, we retain two decimal place reporting.

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AMDEL CORE SERVICES  
High Pressure Liquid Analysis - Method GL-02-03  
Page 1 of 3

Client: SANTOS Ltd. Report # 009/354  
Sample: KOREE 1, DST-1  
Separator: 340 psi  
Date: 10/4/90, Cylinder #EX244

After compression to a pressure greater than that at  
which it was collected, the sample was flashed to  
atmosphere at 25°C.

The volume of each is then measured and the density  
of the oil determined. Both fractions are then  
analysed chromatographically. Mathematical  
recombination of the results gives an analysis of the  
sample as received.

The sample yielded 381 mls of liquid  
and 20.0 litres of gas at 25.1 °C.

Flash Gas Analysis

Component Gas	Mol (%)
Nitrogen	0.43
Carbon Dioxide	6.70
Methane	23.64
Ethane	18.39
Propane	26.39
I-Butane	3.03
N-Butane	11.85
I-Pentane	3.19
N-Pentane	3.40
Hexanes	2.20
Heptanes	0.68
Octanes and Higher Hydrocarbons	0.10
Total	100.00
Calculated Gas Density (rel air =1)	1.393

Calorific Value (15.0 deg C, 101.325 kPa)

Gross:	2130.24 BTU/CU Ft	79.35 MJ/CU.M
Nett:	1956.87 BTU/CU Ft	72.89 MJ/CU.M
Gross calorific value of water-satura		77.99 MJ/CU.M
Average Molecular Weight =	40.126	

This report relates specifically to the sample  
tested; it also relates to the batch insofar as the  
sample is representative of the Batch.

Approved Signatory R. Tanke

Date 16-Jun-90

AMDEL CORE SERVICES  
High Pressure Liquid Analysis -Method GL-02-03  
Flash Liquid Analysis

NATA Certificate

Page 2 of 3

Client: SANTOS Ltd.

Report # 009/354

Sample: KOREE 1, DST-1  
Separator: 340 psi  
Date: 10/4/90, Cylinder #EX244

16-Jun-90

Boiling Point Range (Deg.C)	Component	Weight%	Mol%
-88.6	Ethane	0.06	0.20
-42.1	Propane	1.00	2.31
-11.7	I-Butane	1.35	2.36
-0.5	N-Butane	3.50	6.12
27.9	I-Pentane	4.75	6.69
36.1	N-Pentane	8.10	11.41
36.1-68.9	C-6	9.60	11.32
80.0	Benzene	0.00	0.00
68.9-98.3	C-7	16.76	17.00
100.9	Methylcyclohexane	8.21	8.50
110.6	Toluene	0.20	0.22
98.3-125.6	C-8	12.76	11.35
136.1-144.4	Ethylbenz+Xylenes	2.48	2.37
125.6-150.6	C-9	7.85	6.25
150.6-173.9	C-10	7.32	5.23
173.9-196.1	C-11	4.13	2.69
196.1-215.0	C-12	3.06	1.83
215.0-235.0	C-13	2.43	1.34
235.0-252.2	C-14	1.60	0.82
252.2-270.6	C-15	1.36	0.65
270.6-287.8	C-16	0.80	0.36
287.8-302.8	C-17	0.54	0.23
302.8-317.2	C-18	0.61	0.24
317.2-330.0	C-19	0.42	0.16
330.0-344.4	C-20	0.29	0.10
344.4-357.2	C-21	0.26	0.09
357.2-369.4	C-22	0.22	0.07
369.4-380.0	C-23	0.11	0.03
380.0-391.1	C-24	0.07	0.02
391.1-401.7	C-25	0.08	0.02
401.7-412.2	C-26	0.07	0.02
412.2-422.2	C-27	0.01	0.00
>422.2	C-28+	0.00	0.00
Total		100.00	100.00
( 0.00 = LESS THAN 0.01% )			

The above boiling point ranges refer to the normal paraffin hydrocarbon boiling in that range. Aromatics, branched hydrocarbons, naphthenes and olefins may have higher or lower carbon numbers but are grouped and reported according to their boiling points.

## Oil Parameters:

Specific Gravity @ 23.0 °C 0.7135  
Specific Gravity @ 15.6 °C 0.7208  
API Gravity 64.81  
Specific Gravity of C8+ fraction 0.8025 (calc)  
Average molecular weight of C-8+ fraction 139

AMDEL CORE SERVICES  
 High Pressure Liquid Analysis - Method GL-02-03  
 Calculated Composition of Total Sample as Received

NATA Certificate

Page 3 of 3

Client: SANTOS Ltd.

Report # 009/354

Sample: KOREE 1, DST-1

16-Jun-90

 Separator: 340 psi  
 Date: 10/4/90, Cylinder #EX244

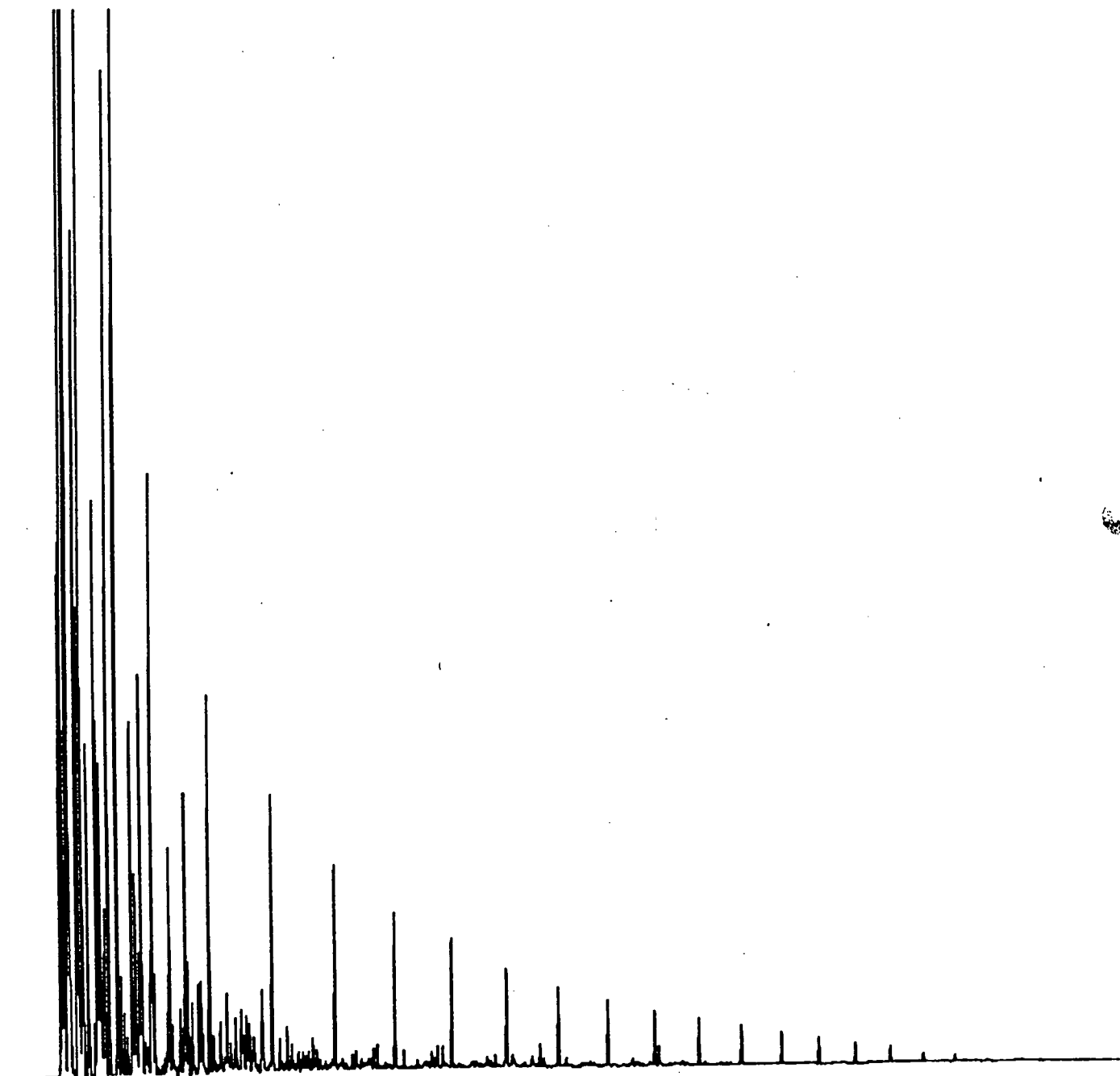
Boiling Point Range (Deg.C)	Component	Weight%	Mol%
-195.8	Nitrogen	0.03	0.10
-78.5	Carbon Dioxide	0.83	1.63
-164	Methane	1.06	5.74
-88.6	Ethane	1.60	4.62
-42.1	Propane	4.21	8.28
-11.7	I-Butane	1.69	2.52
-0.5	N-Butane	5.04	7.50
27.9	I-Pentane	4.86	5.83
36.1	N-Pentane	7.87	9.45
36.1-68.9	C-6	9.05	9.09
80.0	Benzene	0.00	0.00
68.9-98.3	C-7	15.05	13.02
100.9	Methylcyclohexane	7.28	6.42
110.6	Toluene	0.18	0.17
98.3-125.6	C-8	11.35	8.61
136.1-144.4	Ethylbenz+Xylenes	2.20	1.79
125.6-150.6	C-9	6.96	4.70
150.6-173.9	C-10	6.49	3.95
173.9-196.1	C-11	3.66	2.06
196.1-215.0	C-12	2.71	1.38
215.0-235.0	C-13	2.16	1.01
235.0-252.2	C-14	1.42	0.62
252.2-270.6	C-15	1.21	0.49
270.6-287.8	C-16	0.71	0.27
287.8-302.8	C-17	0.48	0.17
302.8-317.2	C-18	0.54	0.18
317.2-330.0	C-19	0.37	0.12
330.0-344.4	C-20	0.26	0.08
344.4-357.2	C-21	0.23	0.07
357.2-369.4	C-22	0.20	0.05
369.4-380.0	C-23	0.10	0.03
380.0-391.1	C-24	0.06	0.02
391.1-401.7	C-25	0.07	0.02
401.7-412.2	C-26	0.06	0.01
412.2-422.2	C-27	0.01	0.00
>422.2	C-28+	0.00	0.00
	Total	100.00	100.00

00014

KOREE 1 DST 1

10/5/90

009/354



KOREE 1, DST-1-4  
Separator: 340 psi  
Date: 10/4/90

\*\*\*\*\* Flash Gas Flash Liquid  
ooooo HP Gas Recombined Liquid

