

SECTION



G E O L O G I C A L S U R V E Y

Elliston Town Water Supply
Report on Drilling and Pump Testing

by

A.F. WILLIAMS

Department of Mines
South Australia —

73/158

DEPARTMENT OF MINES
SOUTH AUSTRALIA

GEOLOGICAL SURVEY
ENGINEERING DIVISION

ELLISTON TOWN WATER SUPPLY
REPORT ON DRILLING AND PUMP TESTING

by

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GEOLOGIST
HYDROGEOLOGY SECTION

27th June, 1973

Rept.	Bk.	No.	73/158
G.S.	No.	5163	
Hyd.	No.	2556	
D.M.	No.	1015/69	

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DEPARTMENT OF MINES
SOUTH AUSTRALIA

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ELLISTON TOWN WATER SUPPLY
REPORT ON DRILLING AND PUMP TESTING

INTRODUCTION

At the request of the Engineering and Water Supply Department, a bore hole was drilled and pump tested by the Drilling Branch of the Mines Department to provide an alternate source for the town water supply (see Fig. 1). This bore is some 0.8 km ($\frac{1}{2}$ m) from the main producing bore. Mr. W.H. Deer, Resident Engineer for the above Department at Port Lincoln indicated that a supply of about 6 litres per second (4 500 gallons per hour) should be sufficient for an auxiliary source.

RESULTS

This bore intersected some 84m of calcarenite (see logg appendix 1) which was generally poorly cemented but some harder bands were found at depth. The whole of the sequence penetrated is considered part of the Bridgewater Formation. The bore was subsequently pump tested with a line shaft pump. An airline and pressure gauge arrangement was used to record the pressure decrease in the hole during the test (each 1 kilopascal pressure decrease corresponded to a decrease of 0.1 metre of drawdown). The results are displayed on Table 1. The maximum drop of pressure recorded during the test was 14 kilopascals (2 lbs/sq.in) which corresponded to 1.4 metres of drawdown. The bore hole developed slightly during the test and at the completion, the pressure had recovered by

7 kilopascals equivalent to a final drawdown of 0.7 metres.

The bore fully recovered in 20 minutes (see Table I). Pumping rate averaged about 11.3 litres per second. Salinity results (about 900 mg/l - see appendix II) indicated that the water is suitable for human consumption.

CONCLUSIONS

This bore should provide an adequate auxiliary source for the Elliston town water supply. The bore can be safely pumped at about 11 litres per second. No problems are anticipated since the supply required is of the order of 6 litres per second.

AFW:jl
27th June, 1973

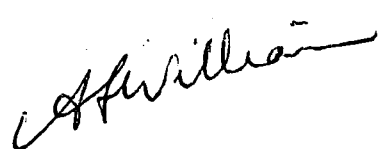

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Table I - showing time, pressure readings and pump rates.

Time (minutes after start)	Pressure reading (kilopascals)	Pumping rate litres per second
0	179	-
303	176	10.9
60	172	11.0
9090	169	11.1
120	165	11.2
150	165	"
180	165	"
720	165	11.2
1350	165	11.2
1980	165	11.4
2370	169	11.1
2460	171	11.1
2700	171	11.2
1790	171	11.3
Pump stopped at 2790 minutes after start Bore then allowed to recover		
2880	172	
2890	191	Release of water from pump column?
2900	179	- Recovery complete

Note (1) 1 kilopascal = 0.145 pounds per square inch.
(ii) Average pump rate 11.3 litres per second.

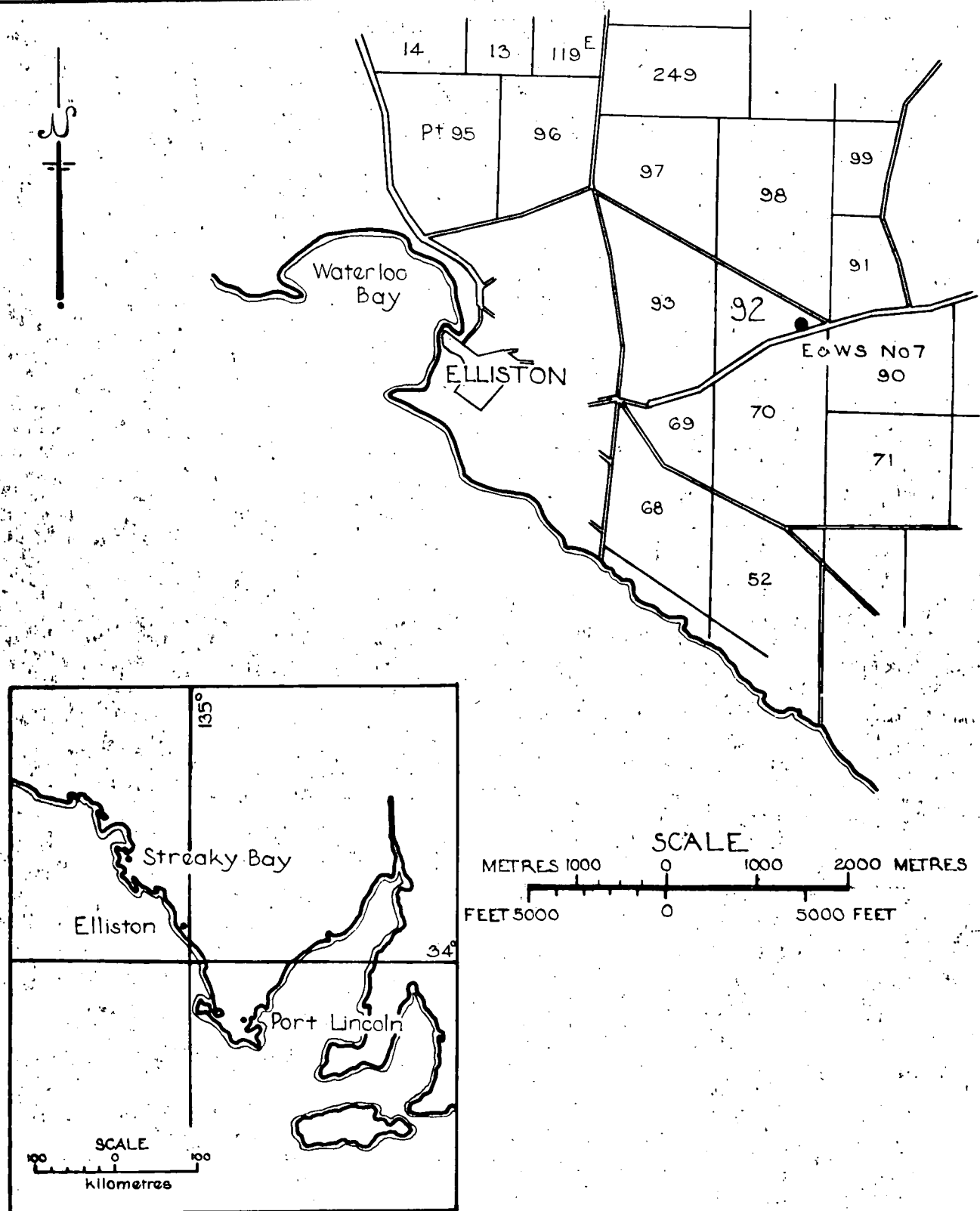


FIG. 1

DEPARTMENT OF MINES – SOUTH AUSTRALIA

Scale: 1:63 360

Compiled: A.F.W.

**LOCALITY PLAN
ELLISTON TOWN WATER SUPPLY
AUXILIARY BORE**

Date: 27 June 1973

Drn. R.B. Ckd.

Drg. No.
S10361
D.L.G.

Purpose of Bore **ELLISTON TOWN WATER SUPPLY.**

State No. 680009203

Hundred WARD.

Section . 92

Bore Serial No. 32/73

Owner, E. & W. S.

Address VIET

IDE Project No. BORE 6

Driller, A. STURAK

• • •

Docket No. 1015/69

Commenced. 19.3.73

Completed 30.3.73

R.L. Collar (M.S.L.)

Depth 84 m

Drill type **CABLE TOOL** Circulation

R.L. Surface

Co-ords E

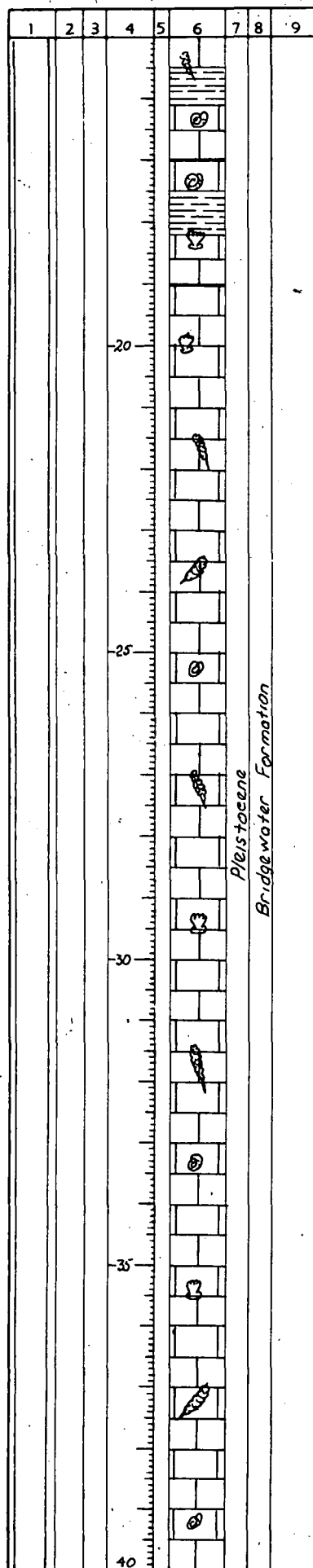
Logged by A. F. WILLIAMS Date 8.5.73

Casing see log A.M.G. Zone

2

REMARKS:

CASING	WATERS CUT	WATER LEVEL	DEPTH (m)	CORE	GRAPHIC LOG	AGE	UNIT	PENETRATION RATE	DESCRIPTION
80.54m of 6" Casing			0						0 - .84 m <u>Calcarenite</u> 95% carbonate, 5% quartz fine to medium grained, subangular to well rounded, possibly some oolitic grains, pink, grey, white and yellow. Poorly cemented. Possible shell fragments
			5						
			10						11-15 m 5% clay.
			15						



15-17 m as above 10-15% clay.

17-19 m as above 5-10% quartz, less clay. Some carbonate cement.

19-25 m as above but some harder material at 19-21 m.

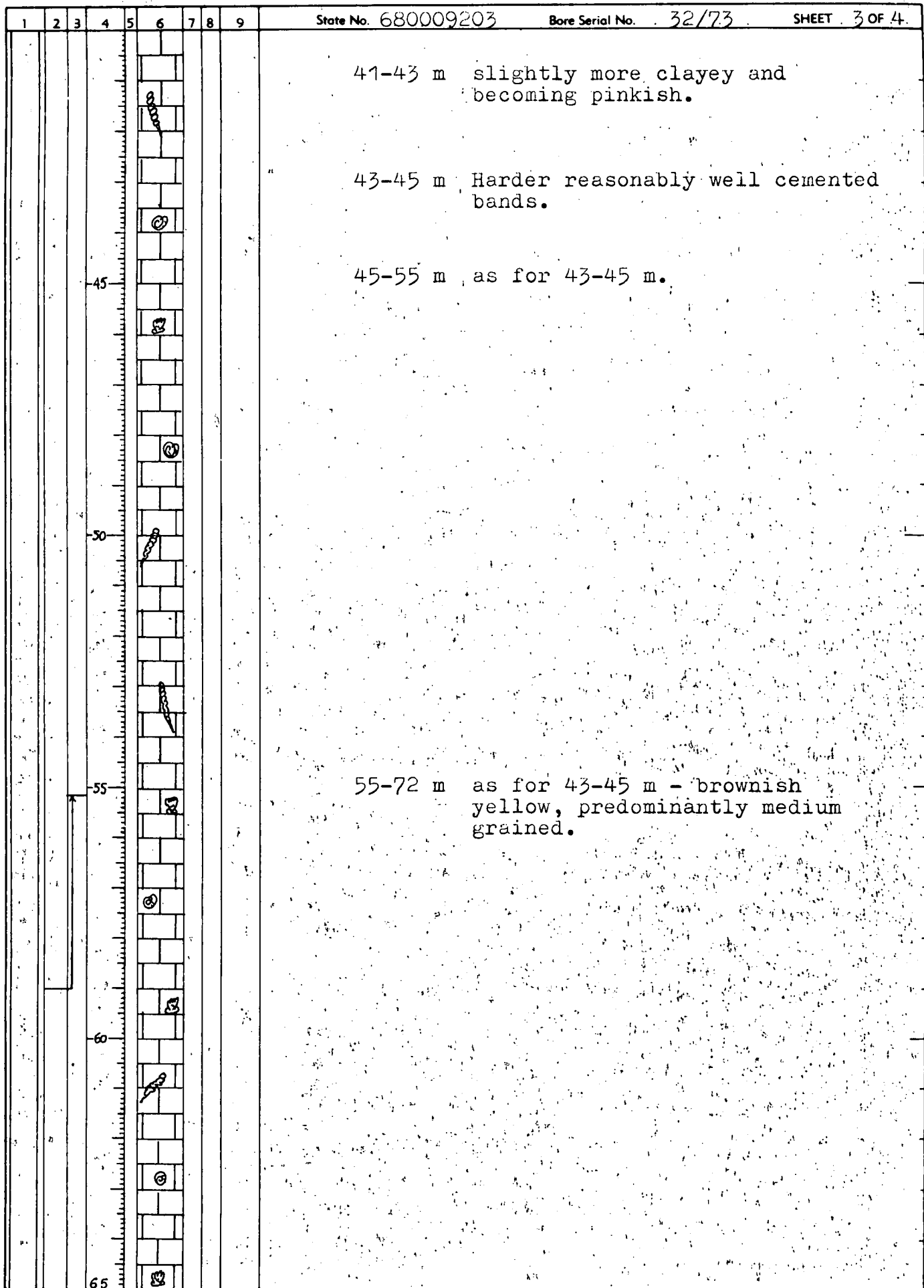
25-41 m as above but becoming more yellowish.

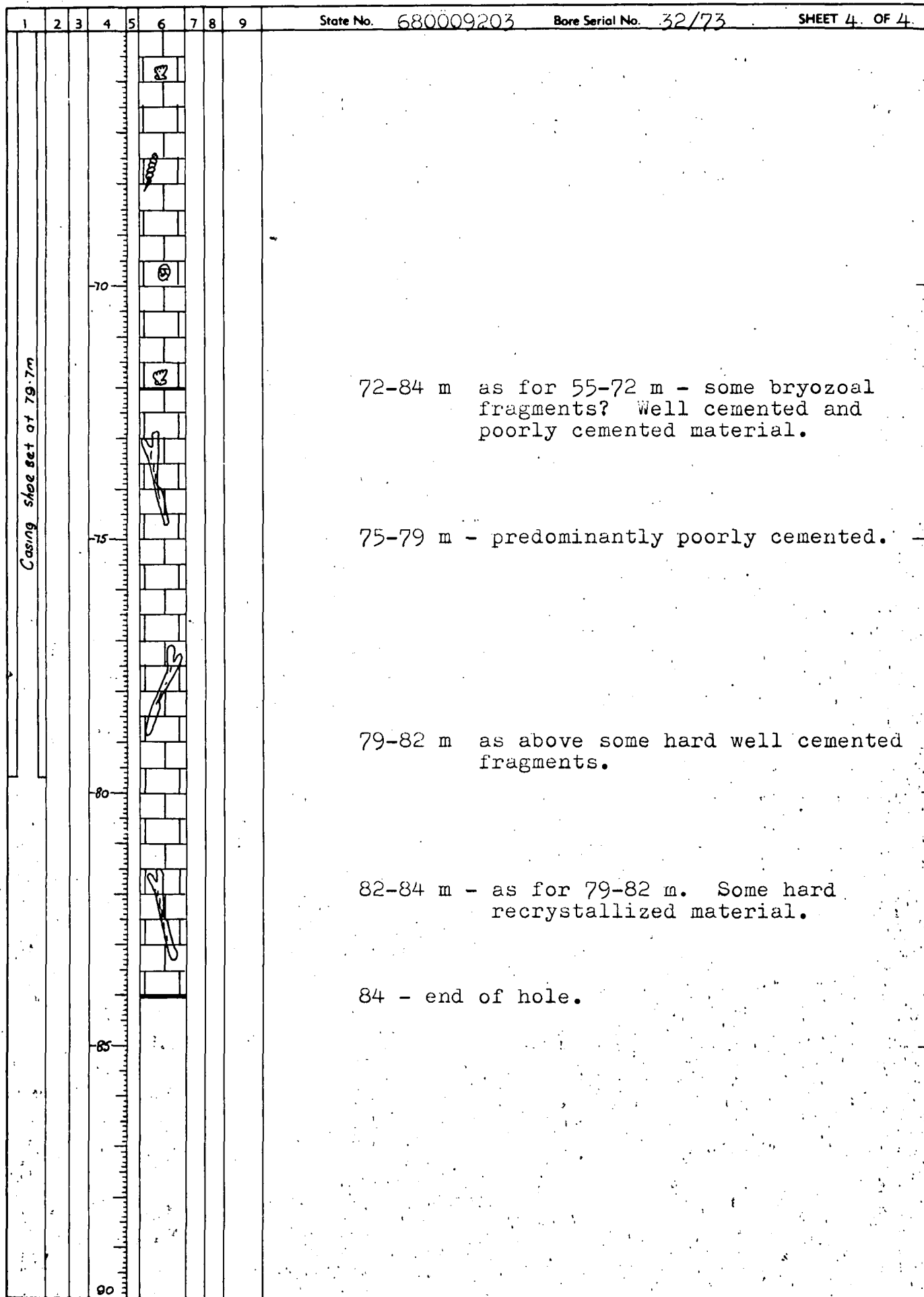
41-43 m slightly more clayey and becoming pinkish.

43-45 m Harder reasonably well cemented bands.

45-55 m, as for 43-45 m.

55-72 m as for 43-45 m - brownish yellow, predominantly medium grained.





WATER ANALYSIS REPORT

SAMPLE NO. W1692/73

JOB NO. 4328/73

CHEMICAL COMPOSITION

		Milligrams per litre mg/l	Milliequivalents per litre me/l
Cations			
Calcium	(Ca)	<u>69</u>	<u>3.4</u>
Magnesium	(Mg)	<u>45</u>	<u>3.7</u>
Sodium	(Na)	<u>200</u>	<u>8.7</u>
Potassium	(K)	<u>7</u>	<u>0.2</u>
Iron	(Fe)		
Anions			
Carbonate	(CO ₃)		
Bicarbonate	(HCO ₃)	<u>305</u>	<u>5.0</u>
Sulphate	(SO ₄)	<u>50</u>	<u>1.0</u>
Chloride	(Cl)	<u>358</u>	<u>10.1</u>
Fluoride	(F)		
Nitrate	(NO ₃)	<u>7</u>	<u>0.1</u>
Phosphate	(PO ₄)		

TOTALS and BALANCE

Cations me/l	Anion me/l
<u>16.0</u>	<u>16.2</u>
diff Δ <u>0.2</u>	sum Σ <u>32.2</u>
% (Δ x 100)	<u>0.62</u>

DERIVED AND OTHER DATA

Conductivity (E.C.) μS/cm at 25°C	<u>1705</u>
Total Dissolved Solids:	
a. Based on E.C.	
b. Calculated (HCO ₃ =CO ₃)	<u>886</u>
c. Residue on evaporation at 180°C	
Total Hardness as CaCO ₃	<u>355</u>
Carbonate Hardness as CaCO ₃	<u>250</u>
Non-carbonate Hardness as CaCO ₃	<u>105</u>
Total Alkalinity as CaCO ₃	<u>250</u>
Free Carbon Dioxide (CO ₂)	
Suspended Solids	
Silica (SiO ₂)	
Boron (B)	
Reaction-pH	
Turbidity (Jackson)	
Colour (Hazen)	
Sodium Absorption Ratio	<u>54.4</u>
Sodium to total cation ratio	

Name E. W. S. Hundred Ward Supply 4,000

Section 680009203 Depth Hole 84.00

Address Ellistontown Hole No. 6 Date Collected 28.3.73

Water Supply Water Cut 59.00 Sample Collected by A Sturak

Water Level 55.15 Date Received

REMARKS:

WATER ANALYSIS REPORT

SAMPLE NO. **W1265/73**

JOB NO. **4483/73**

CHEMICAL COMPOSITION

Milligrams
per litre
mg/l

Milliequivalents
per litre
me/l

Cations

Calcium	(Ca)	57	2.8
Magnesium	(Mg)	49	4.0
Sodium	(Na)	218	9.5
Potassium	(K)	8	0.2
Iron	(Fe)		

Anions

Carbonate	(CO ₃)		
Bicarbonate	(HCO ₃)	310	5.1
Sulphate	(SO ₄)	50	1.0
Chloride	(Cl)	365	10.3
Fluoride	(F)		
Nitrate	(NO ₃)	8	0.1
Phosphate	(PO ₄)	0.08	

TOTALS and BALANCE

Cations me/l **16.5** Anion me/l **16.5**

diff Δ **0** sum Σ **33.0** % ($\frac{\Delta \times 100}{\Sigma}$) **-**

DERIVED AND OTHER DATA

Conductivity (E.C.) μS/cm at 25°C **1945**

Milligrams
per litre
mg/l

Total Dissolved Solids:

- a. Based on E.C. **908**
- b. Calculated (HCO₃=CO₃) **908**
- c. Residue on evaporation at 180°C **908**

Total Hardness as CaCO₃ **345**

Carbonate Hardness as CaCO₃ **255**

Non-carbonate Hardness as CaCO₃ **90**

Total Alkalinity as CaCO₃ **255**

Free Carbon Dioxide (CO₂)

Suspended Solids

Silica (SiO₂)

Boron (B)

Units

Reaction—pH

Turbidity (Jackson)

Colour (Hazen)

Sodium Absorption Ratio
Sodium to total cation ratio
(equiva) **57.6**

Name **Elliston Town Supply** Hundred **Ward** Supply **8960 gph**

Section **92** Depth Hole

Address Hole No. **7** Date Collected **11.4.73**

Water Cut Sample Collected by **R. Tait**

Water Level Date Received

REMARKS:

for Director